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13. SUPPLEMENTARY NOTES

AFOSR Ionic Liquids Workshop

Tampa, FL. 7-8 March 2004 14. ABSTRACT

15. SUBJECT TERMS

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# IL QC QSPR - Preliminary Results

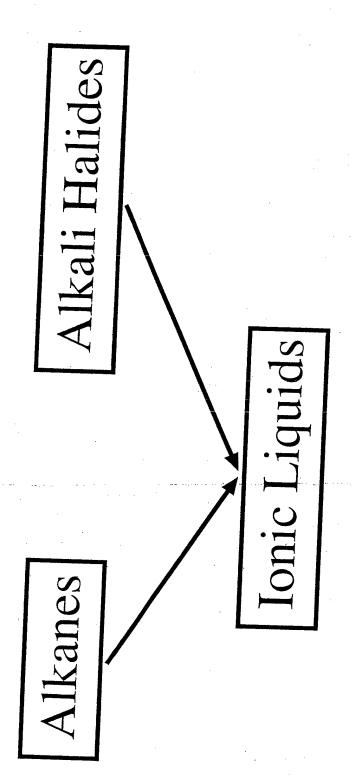
Jeffrey D. Mills Jeffrey.Mills@edwards.af.mil

Space and Missile Propulsion Division Air Force Research Laboratory Propellants Branch AFRL/PRSP

### QSPR



Quantitative Structure-Property Relationships Property(Descriptor)



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### Property (Descriptor)



Properties:

(Impact Sensitivity) (Decomp. Temp.) (Density (Solid)) Density (liquid) Melting Point (Others??)

QC Descriptors:

- Single Particle:

Surface Size and Shape Surface Electrostatic Potential

- Pair:

Binding Energy Separation





### Constraints/Goals



1. Predictive (not just summarize or interpolate)

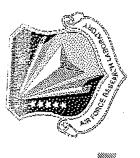
"Universal" Descriptors (Ionic and Nonionic)

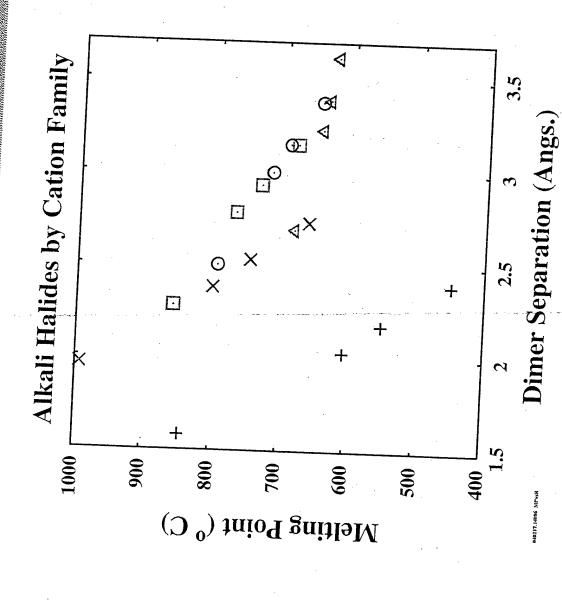
3. Allow Ion Interchangablility

4. Physically Meaningful and Chemically Ignorant a. No "Kitchen-Sink" Fits

b. Charge Symmetry

## Cation Families Summarized



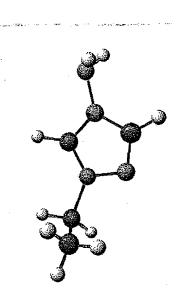


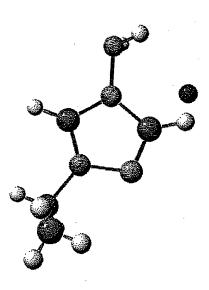
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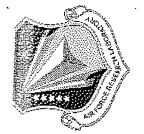
B3LYP 6-311++G(d,p): Crystal (with Br<sup>-</sup>): 1-Ethyl-4-amino-1,2,4-triazolium Cation











Color Map:

V at 0.001 au Electron Density:

kcal/mol

350.0

291.7

Na+

233,3

 $\mathbb{R}b^+$ :

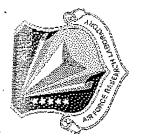
175.0





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+1/1

| |---

kcal/mol

350.0

250.0

200.0

150.0

100.0

0.0

50.0

-50.0

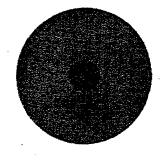
-150.0

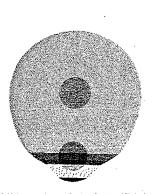
-200.0

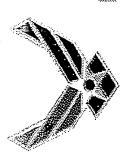
-250.0

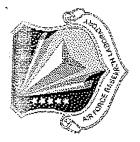
**-350.0** 

-300.0

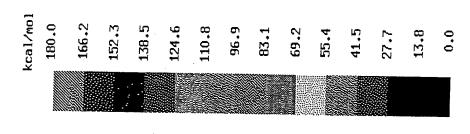


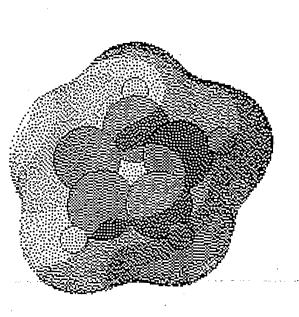


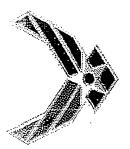




### 1,2,4-Triazolium Cation









# 1-R<sub>n</sub>-4-amino-1,2,4-triazolium Cation, n=even:

n=0:

n=6:

kcal/mol

166.2

152.3

124.6

n=2:

110.8

6.96

83.1

55.4

41.5

n=4:



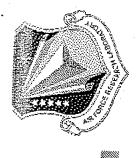
n=8:







### Descriptors



Electrostatic Potential:

$$\Pi = rac{1}{\mathrm{A}} \sum_{\mathrm{i}} |\mathrm{V_i} - ar{\mathrm{V}}|\,\mathrm{A_i}$$

Size

$$\mathrm{A} = \sum_{\mathbf{i}} \mathrm{A}_{\mathbf{i}}$$

Shape:

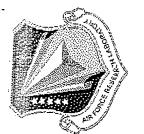
$$(I_0, I_1, I_2)$$
 - PA Moments of Extensia (ordered)

Asphericity = 
$$\frac{I_2 - I_0}{\bar{\tau}}$$

Blateness = 
$$\frac{I_1 - I_0}{I_2 - I_0}$$

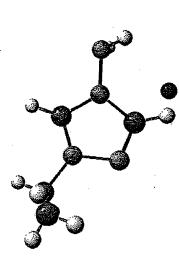


# Separation and Interaction Energy

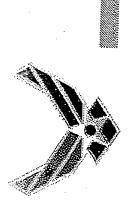


1-Ethyl-4-amino-1,2,4-triazolium Bromide B3LYP 6-31+G(d):

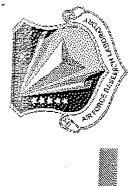
Crystal:

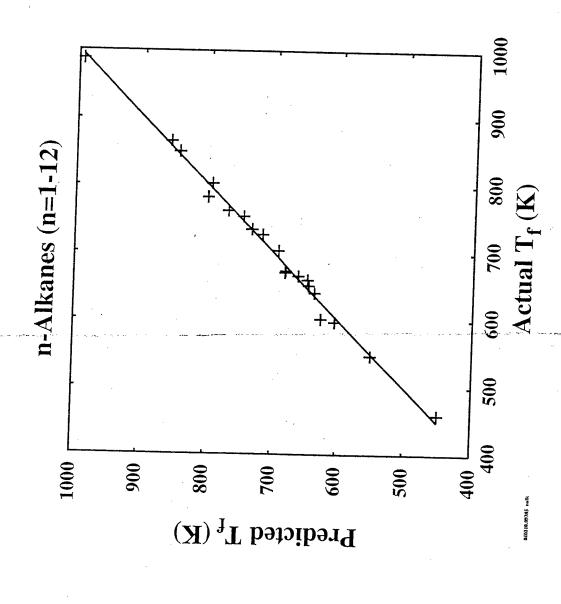


Binding Energy

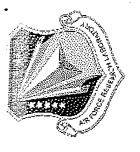


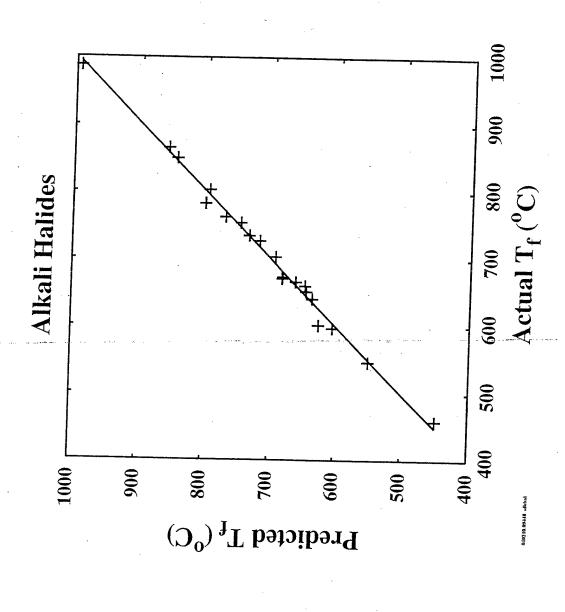
### QSPR Correlation





### QSPR Correlation



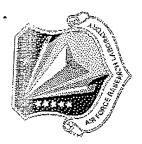


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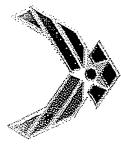




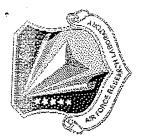
## Concluding (Open) Question



Can QC QSPR aid the IL synthesist or will it merely follow?



### Acknowledgements



Peter Politzer

J. Boatz

DoD HPCMP - ASC

G. Drake, L. Hall